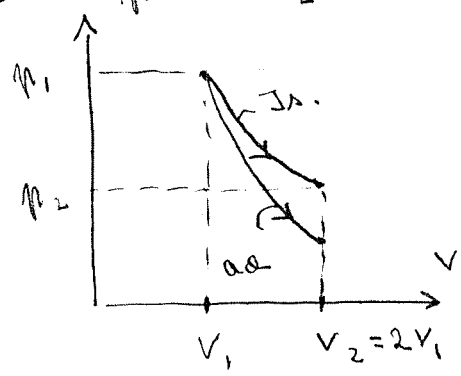


Oppg 1. Løsning

a) $W = \int_{V_1}^{2V_1} p \, dV = n_1 V_1 \int_{V_1}^{2V_1} \frac{dV}{V} = n_1 V_1 \ln 2 \approx 0,7 n_1 V_1$

b) $W_{ad} = n_1 V_1^{5/3} \int_{V_1}^{2V_1} \frac{dV}{V^{5/3}} = \frac{3}{2} n_1 V_1^{5/3} \left[-\frac{1}{V^{2/3}} \right]_{V_1}^{2V_1}$
 $= \frac{3}{2} n_1 V_1 \left[1 - \frac{1}{2^{2/3}} \right] \approx 0,55 n_1 V_1$

c) [alt: $Q=0 \Rightarrow W_{ad} = -\Delta U = U_1 - U_2$]



areal under isoterm >
areal under adiabat

$W > W_{ad}$ som over

d) Trykkel: Fra molekylene stot mot veggen
 Temp: Mal for middlere kinetiske molekylene energi

Isoterm: Konst T \Rightarrow konst $\langle K \rangle \Rightarrow$ konst fart

Trykkel antar fordi molekylene stoter sjeldnere mot veggen nar V oker

Adiabat: Utfort arbeid n_1 bevarer energi av indre energi: K antar \Rightarrow fart antar

Trykkel antar av samme grunn som overfor. I tillegg mindre impuls til veggen ved hvert stot

~~Omgå a~~

$$a) N_1/N = C e^{-\frac{\epsilon}{\theta}} = C \quad N_2/N = C e^{-\frac{\epsilon}{\theta}} = C e^{-\frac{\epsilon}{\theta}}$$

$$1 = N_1/N + N_2/N \Rightarrow C^{-1} = 1 + e^{-\frac{\epsilon}{\theta}} \quad \therefore$$

$$N_1 = \frac{N}{1 + e^{-\frac{\epsilon}{\theta}}}$$

$$N_2 = \frac{N}{1 + e^{\frac{\epsilon}{\theta}}}$$

$$b) \bar{\epsilon} = \frac{N_1}{N} \frac{\epsilon}{2} + \frac{N_2}{N} \frac{\epsilon}{2} \Rightarrow \bar{\epsilon} = \frac{\epsilon}{1 + e^{\frac{\epsilon}{\theta}}}$$

$$U = N \bar{\epsilon} = \frac{N \epsilon}{1 + e^{\frac{\epsilon}{\theta}}}$$

$$C_v = \frac{\partial U}{\partial T} = k \frac{\partial U}{\partial \theta} \Rightarrow C_v = N k \left(\frac{\epsilon}{\theta}\right)^2 \frac{e^{\frac{\epsilon}{\theta}}}{(1 + e^{\frac{\epsilon}{\theta}})^2}$$

$$c) \theta \gg \epsilon \Rightarrow e^{\frac{\epsilon}{\theta}} \approx 1 \quad e^{-\frac{\epsilon}{\theta}} \approx 1 \Rightarrow$$

$$N_1 \approx N/2, \quad N_2 \approx N/2, \quad U \approx N \epsilon / 2, \quad C_v \approx \frac{1}{4} N k \left(\frac{\epsilon}{\theta}\right)^2$$

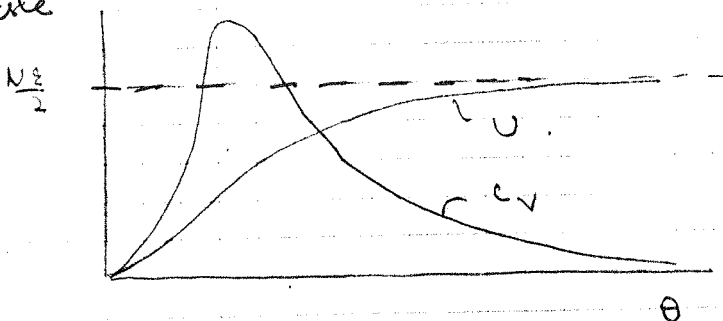
Likefordeling

$$\theta \ll \epsilon \Rightarrow e^{\frac{\epsilon}{\theta}} \gg 1 \quad e^{-\frac{\epsilon}{\theta}} \ll 1 \Rightarrow$$

$$N_1 \approx N, \quad N_2 \approx N e^{-\frac{\epsilon}{\theta}} \approx 0$$

$$U \approx N \epsilon e^{-\frac{\epsilon}{\theta}} \approx 0 \quad C_v \approx N k \left(\frac{\epsilon}{\theta}\right)^2 e^{-\frac{\epsilon}{\theta}}$$

Alle i laveste nivå



Öppg. 3. Lös

3

a)

$$1. \text{ HS} \quad \dot{q}_K + W = \dot{q}_V$$

$$2. \text{ HS} \quad -\dot{q}_K/T_K + \dot{q}_V/T_V \geq 0$$

$$\dot{q}_V \geq \dot{q}_K \frac{T_V}{T_K} \Rightarrow W > \dot{q}_K \frac{T_V}{T_K} - \dot{q}_K$$

$$W_{\min} = \dot{q}_K \left(\frac{T_V}{T_K} - 1 \right) = \dot{q}_K \frac{T_V - T_K}{T_K}$$

b) ε effektivitet = $\frac{\text{Res.}}{\text{Kostn.}} = \frac{\text{gjennet varme}}{\text{brukt arbeid}}$

$$\varepsilon_{\max} = \frac{\dot{q}_K}{W_{\min}} = \frac{T_K}{T_V - T_K}$$

$$T_K = -15^\circ \text{C} = 258 \text{K}$$

$$T_V = 20^\circ \text{C} = 293 \text{K}$$

$$\varepsilon_{\max} = \frac{258}{293 - 258} = \frac{258}{35} \approx 7,4$$

c) $\varepsilon = \varepsilon_{\max} / 2 = 3,7$

$$W = 250 \cdot 0,15 = 37,5 \text{ W} = 37,5 \text{ J/s}$$

Varme som beholder min m^o (per sec):

$$\dot{q} = \dot{q}_K = \varepsilon W = 3,7 \cdot 37,5 = 138 \text{ J/s}$$

Oppg 4. Løsning

a) Konstant temperatur $T = -13^{\circ}\text{C} = 260\text{K}$

Tilført varme $Q = 3000 \cdot 10^{-3} = 3\text{J}$.

$$\Delta S = \frac{Q}{T} = \frac{3}{260} = \underline{1.15 \cdot 10^{-2}\text{J/K}}$$

$$S = k \ln W \Rightarrow$$

$$S_{\text{fri}} = k \ln W_{\text{fri}} \quad S_{\text{etter}} = k \ln W_{\text{etter}}$$

$$\Delta S = k \ln \frac{W_{\text{etter}}}{W_{\text{fri}}}$$

$$W_{\text{etter}}/W_{\text{fri}} = Q^{\Delta S/k} = 10^{0.43 \Delta S/k} \Rightarrow$$

$$W_{\text{etter}}/W_{\text{fri}} \approx 10^{3.6 \cdot 10^{20}} \gg \gg \gg 1.$$

b)

Alle mikrotilstander forekommer like ofte. Minimal sjans for å finne systemet i en av de W_{fri} tilstandene.

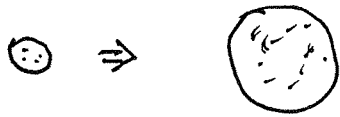
$$\text{Antall molekyler } N = \frac{M}{m_{\text{H}_2\text{O}}} = \frac{10^{-6}}{3 \cdot 10^{-26}} \approx 3 \cdot 10^{19}$$

Antall måter å plassere disse på i flasken \gg antall måter å plassere et snøfnugg på i flasken.

Oppg. 5 løst

5

a)



R_1, T_1

R_2, T_2

Adiabatt $pV^\gamma = \text{konst.}$

Gasslov $pV = NkT$

Kombinasjon

$$(T/V) \cdot V^\gamma = \text{konst.}$$

$$TV^{\gamma-1} = \text{konst.}$$

$$V = \frac{4\pi}{3} R^3 \Rightarrow TR^{3(\gamma-1)} = \text{konst.} \quad \underline{RT^{\frac{1}{3(\gamma-1)}} = \text{konst.}}$$

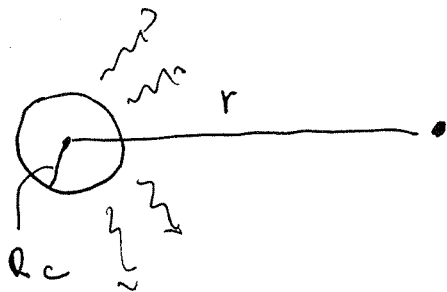
Anvendt på gasskuler:

$$R_2 T_2^{\frac{1}{3(\gamma-1)}} = R_1 T_1^{\frac{1}{3(\gamma-1)}} \quad \underline{R_2 = R_1 \left(\frac{T_1}{T_2} \right)^{\frac{1}{3(\gamma-1)}}}$$

$$\gamma = 1,4 \quad T_1/T_2 = 100 \quad R_1 = 15 \text{ m} \Rightarrow$$

$$\underline{R_2 \approx 700 \text{ m}}$$

b)



Energibevarelse:

$$\underline{I(R_c) \cdot 4\pi R_c^2 = I(r) \cdot 4\pi r^2}$$

Set fra Capella m. sek.

$$R_c = r \left[\frac{I(r)}{I(R_c)} \right]^{1/2} = r \left[\frac{I(r)}{\sigma T_c^4} \right]^{1/2} \Rightarrow$$

$$\underline{R_c \approx 7 \cdot 10^9 \text{ m} \approx 10 R_\odot}$$