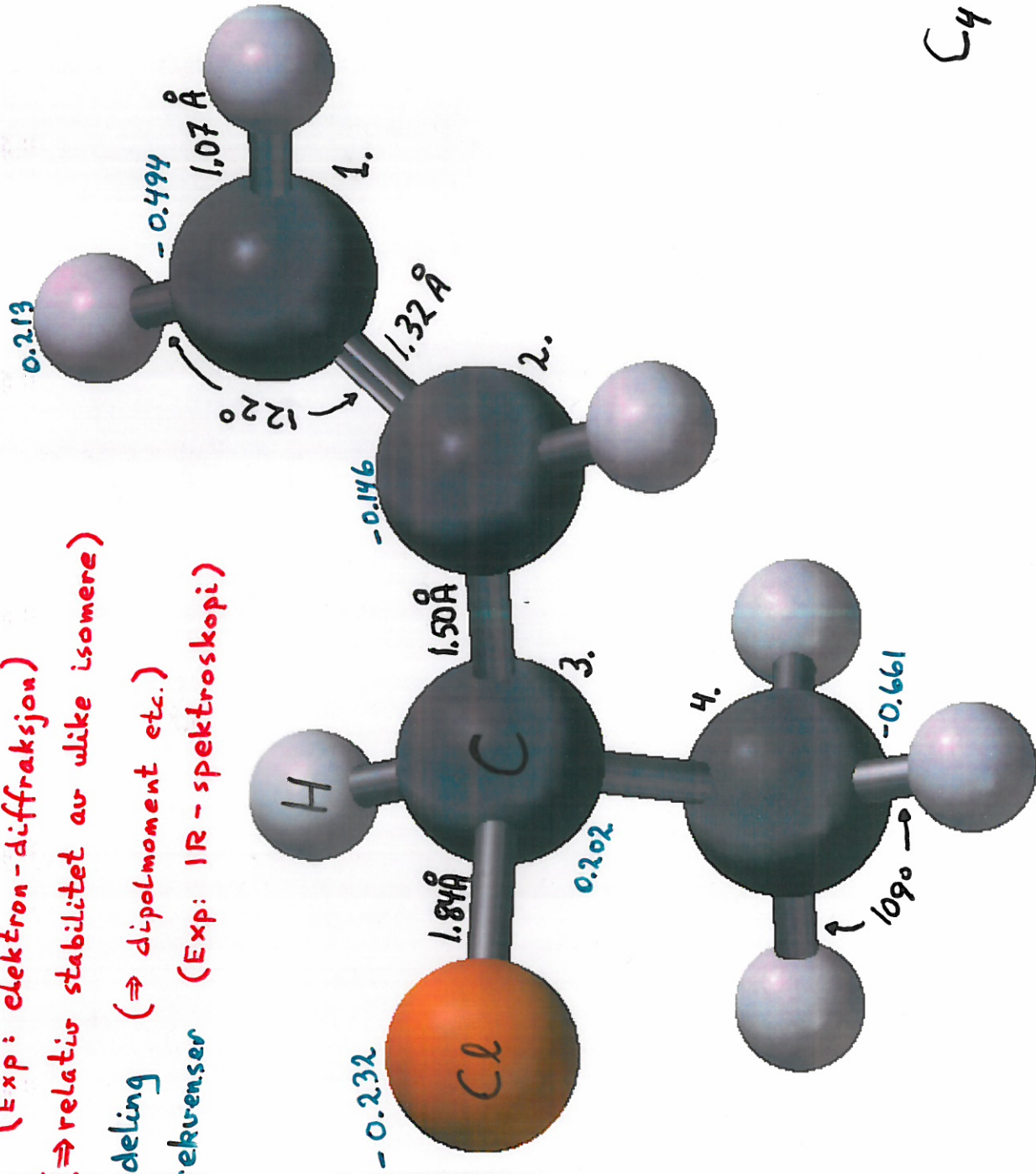


QM beregning $\Rightarrow$

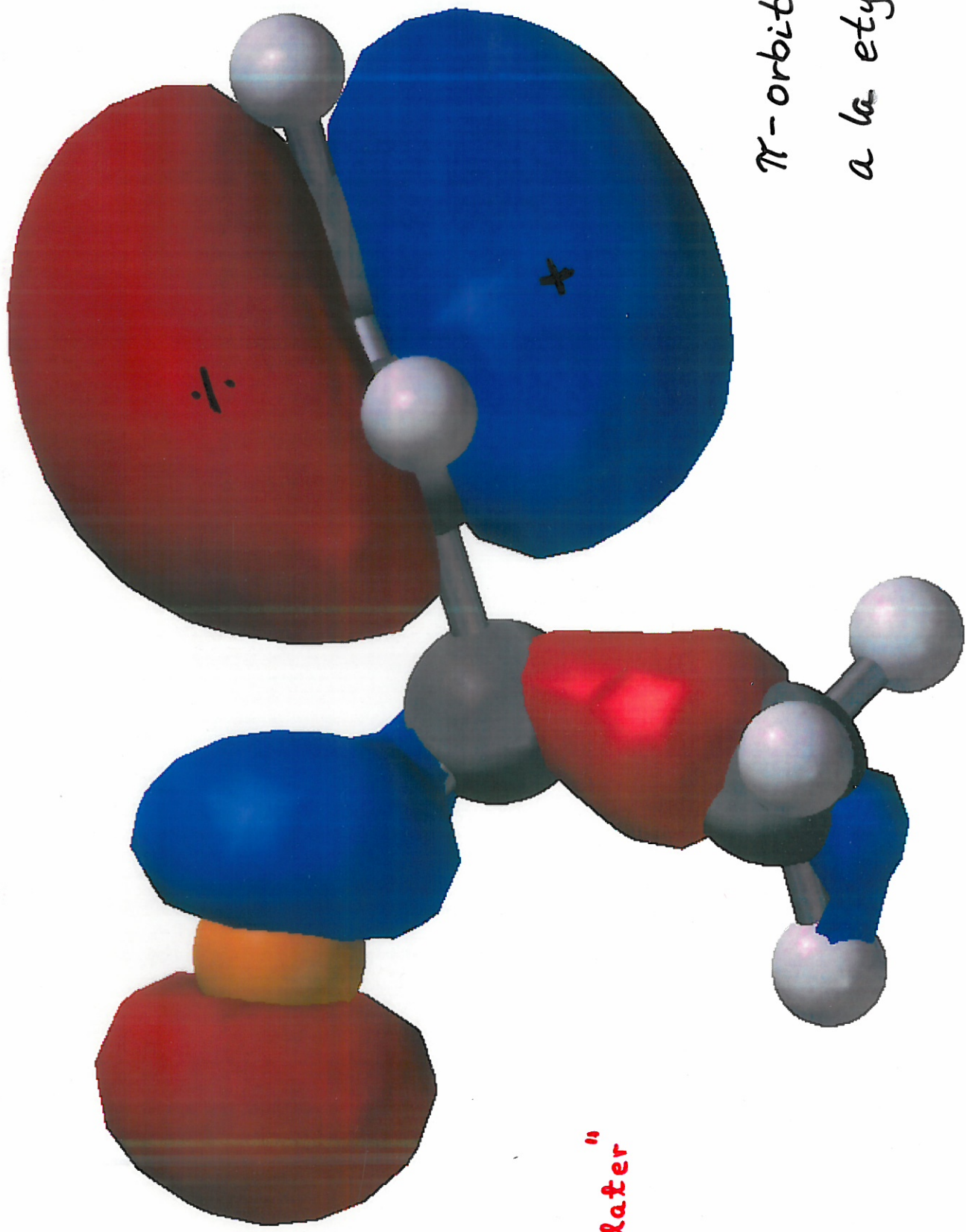
- geometri (Exp: elektron-diffraksjon)
- energi ($\Rightarrow$ relativ stabilitet av ulike isomere)
- ladningsfordeling ($\Rightarrow$ dipolmoment etc.)
- vibrasjons frekvenser (Exp: IR-spektroskopi)



Navn: 3-kloro-1-buten

2.

Molekylorbital

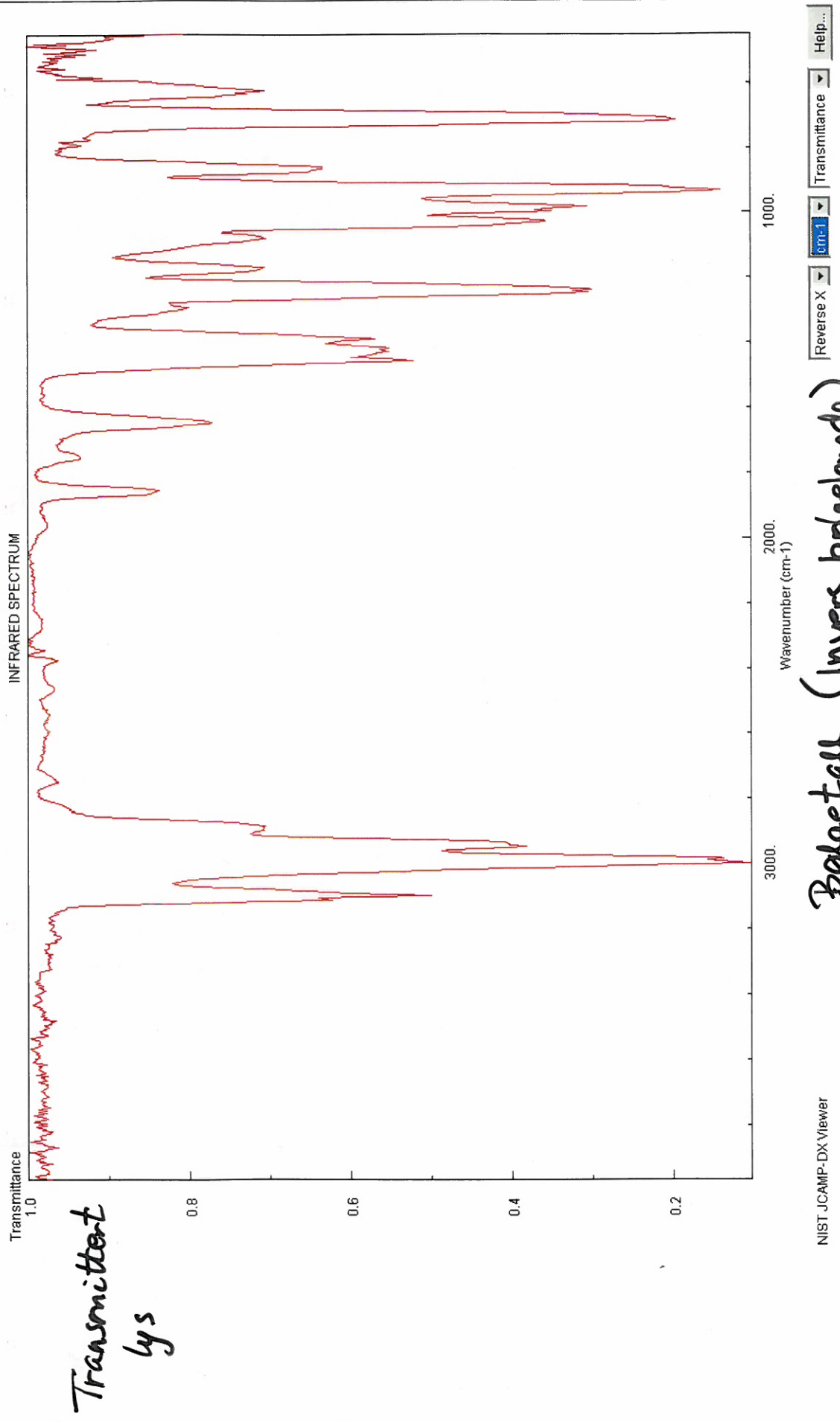


π -orbital
a la etylen

"ekvi- Ψ -flafer"

Absorbere lys med bestemte bølgelængder:

3



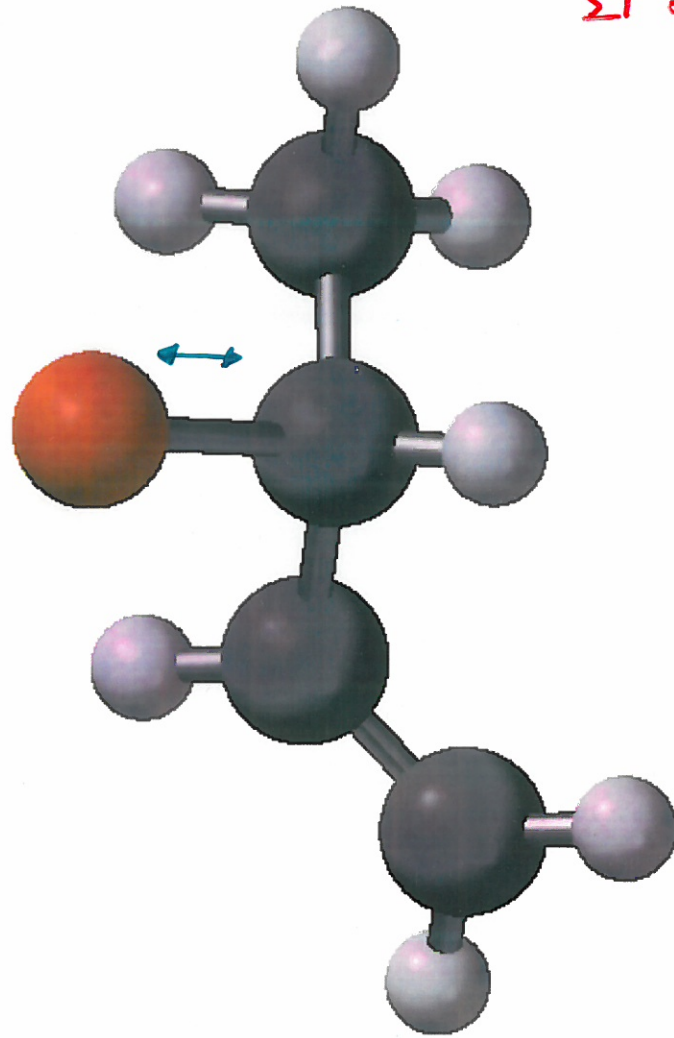
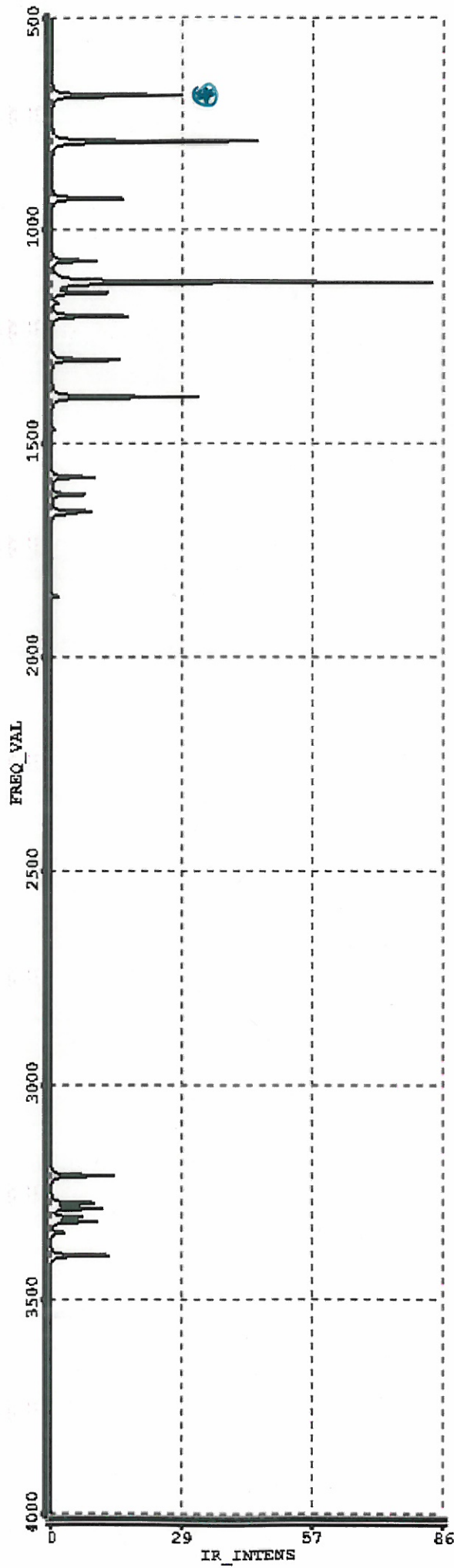
Transmitter
lys

Bølgetal (Invers bølgelængde)

Notice: Concentration information is not available for this spectrum and, therefore, molar absorptivity values cannot be derived.

Beregnet transmisjonsspektrum

4.

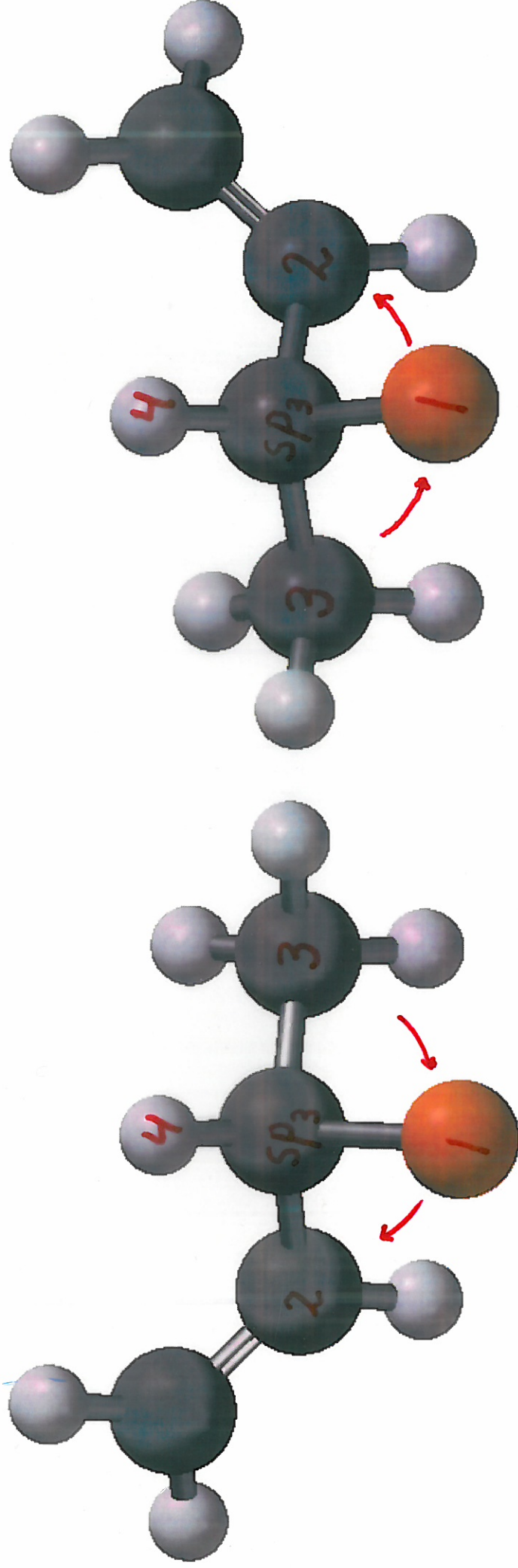


⊕ C-C strekk

Matematisk: Diagonalisering
av matrise; egenverdier
gir frekvensene, egenvektorene
gir vibrasjonsberegningene

Isomere

(= samme formel, ulike molekyler)



hverandres speilbilde!

Ikke identiske egenskaper!

(Roterer polarisasjonsplanet til linearpolarisert lys i hver sin retning $\Rightarrow$ "optiske isomere")